

(19) World Intellectual Property Organization
International Bureau(43) International Publication Date
7 November 2002 (07.11.2002)

PCT

(10) International Publication Number
WO 02/088120 A1

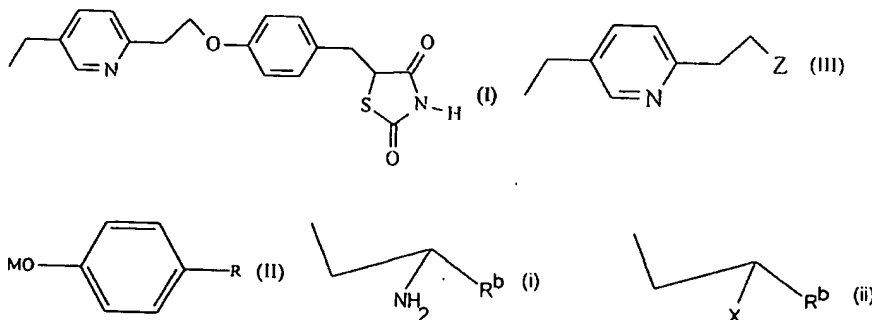
- (51) International Patent Classification⁷: C07D 417/12, A61K 31/4427, A61P 3/10
- (21) International Application Number: PCT/CZ02/00024
- (22) International Filing Date: 25 April 2002 (25.04.2002)
- (25) Filing Language: English
- (26) Publication Language: English
- (30) Priority Data:
PV 2001-1501 26 April 2001 (26.04.2001) CZ
PV 2001-1502 26 April 2001 (26.04.2001) CZ
- (71) Applicant (for all designated States except US): LÉCIVA, A.S. [CZ/CZ]; Dolní Mecholupy 130, 102 37 Praha 10 (CZ).
- (72) Inventors; and
- (75) Inventors/Applicants (for US only): HALAMA, Ales [CZ/CZ]; Lonkova 481, 530 09 Pardubice (CZ). HEJTMÁNKOVÁ, Ludmila [CZ/CZ]; Benesovice 13, 565 01 Písek (CZ). LUSTIG, Petr [CZ/CZ]; Husova 255, 530 03 Pardubice (CZ). RICHTER, Jindrich [CZ/CZ]; Hurka 255, 530 12 Pardubice (CZ). SRSNOVÁ, Lucie [CZ/CZ]; Cs. Armády 471, 537 01 Chrudim (CZ). JIRMAN, Josef [CZ/CZ]; Ceskova 1706, 530 02 Pardubice (CZ).
- (81) Designated States (national): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZM, ZW.
- (84) Designated States (regional): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

Declarations under Rule 4.17:

— as to applicant's entitlement to apply for and be granted a patent (Rule 4.17(ii)) for the following designations AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG,

[Continued on next page]

(54) Title: METHOD FOR OBTAINING PIOGLITAZONE AS AN ANTIDIABETIC AGENT



(57) Abstract: A method for obtaining antidiabetic of formula (I), wherein the method comprises condensing of a 4-derivatized phenol or phenolate of general formula (II), wherein R is an amino group-containing organic residue, selected from the group comprising a residue of the following formula -NHR³, wherein R³ is hydrogen or a protecting group, which is removed before further treatment, and a residue of general formula (A), wherein R^b represents a carboxy group either in the free acid form or in the form of a salt or ester or another functional derivative or the nitrile group CN, and M represents a hydrogen or alkali metal atom, with a pyridine base of general formula (III), wherein Z is a leaving group other than a halogen, wherein, before or after carrying out the condensation, the following operations are carried out: (a) diazotizing the amino group present in organic residue R; (b) converting the diazotised residue R into a derivative of 2-halopropionate or 2-halopropionitrile of formula (B), wherein R^b is as defined above and X is a halogen; (c) cyclizing the derivative of 2-halopropionate or 2-halopropionitrile with thiourea; (d) hydrolysing the resulting imine thus giving pioglitazone of formula (I).



UZ, VN, YU, ZA, ZM, ZW, ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG)

— of inventorship (Rule 4.17(iv)) for US only

— before the expiration of the time limit for amending the claims and to be republished in the event of receipt of amendments

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

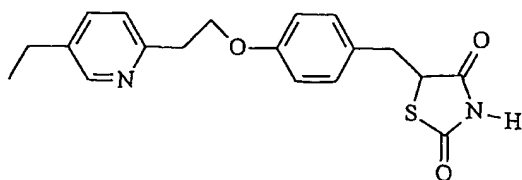
Published:

— with international search report

METHOD FOR OBTAINING PIOGLITAZONE AS AN ANTIDIABETIC AGENT

Technical Field

The present invention relates to a method of obtaining an intermediate, useful for the production of (+/-)-5-[[4-[2-(5-ethyl-2-pyridyl)ethoxy]phenyl] methyl]-2,4-thiazolidinedione (termed pioglitazone hereinafter) of formula I

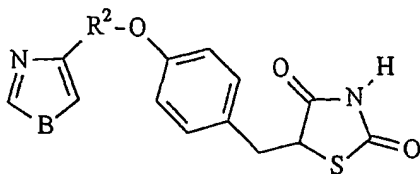


(I)

Pioglitazone belongs to the thiazolidinedione group of antidiabetics. This compound and its antidiabetic properties were described in EP 193 256. Later it was found that its antidiabetic effect consists in reducing insulin resistance, thereby improving glucose availability without increasing insulin secretion, unlike most other antidiabetics. For these extraordinary characteristics this product is of great importance for the treatment of non-insulin dependent diabetes mellitus. Combining with insulin or other antidiabetics can further increase its effect.

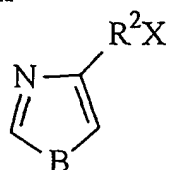
Background Art

Synthesis of thiazolidinediones has been described in EP 177 353. Namely the preparation of compounds of formula VII



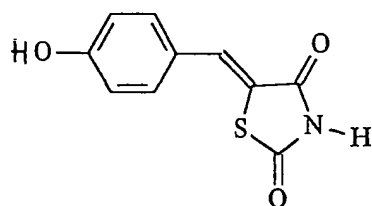
(VII)

wherein B is an oxygen or sulfur atom, R^2 is an alkyl, acyl or hydroxyalkyl, is described by reacting of a compound of formula VIII



(VIII)

wherein X is a halogen,
and a phenolthiazolidinedione of formula VI



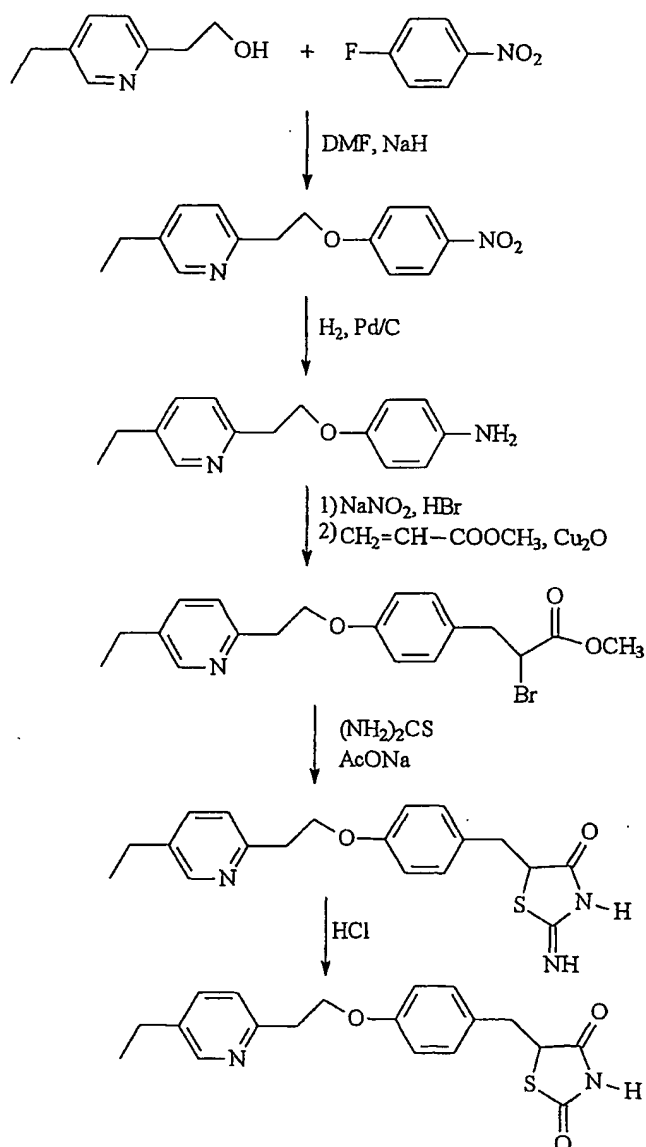
(VI)

The reactions are carried out in alkaline environment in organic solvents (e.g., dimethylformamide).

The method as described in EP 177 353 cannot be simply used for the synthesis of pioglitazone since in reacting 5-ethyl-2-pyridyl-ethylhalide with an appropriate phenol in an alkaline medium elimination reaction resulting in vinyl pyridine will prevail.

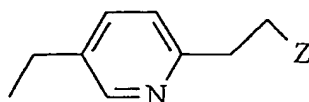
Therefore EP 193 256 discloses another reaction, which can be described by the following scheme

Scheme 1



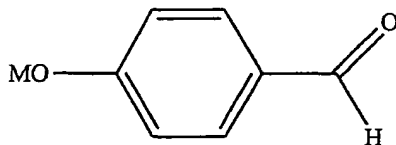
A major drawback of this method is non-standard course of the reduction of the nitro group on palladium (reaction time is 3 hours up to several days), which is apparently dependent on the contents of impurities in the starting compound and in the solvent. The impurities apparently cause poisoning of the catalyst; therefore additional catalyst has to be gradually added. Longer reaction time results in formation of larger amounts of impurities and lower yields.

Further methods how to solve this problem have been elaborated in EP 257 781, in Chem. Pharm. Bull. 39(6) 1440-1445 (1991) and in EP 506 273 and EP 816 340. A common feature of all these methods is the reaction of a compound of formula III



(III)

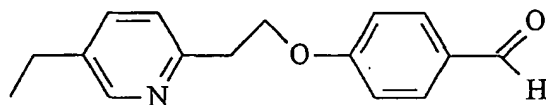
wherein Z is a leaving group of general formula R^3SO_3 wherein R^3 is an alkyl or aryl, with an alkali metal p-hydroxybenzaldehyde or p-formylphenolate, i.e. with a compound of general formula IX



(IX)

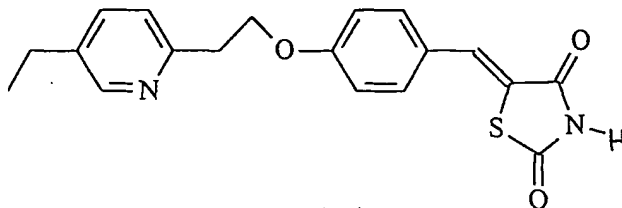
wherein M can represent an alkali metal or hydrogen.

The product of formula X



(X)

further reacts with thiazolidinedione yielding a benzylidene compound of formula XI



(XI)

which is converted to pioglitazone (formula I) by reducing on palladium.

The methods in the above-cited patents differ especially in the media in which reaction of the compound IX with the compound III is carried out. In EP 257 781 and in Chem. Pharm. Bull. 39(6) 1440-1445 (1991) reactions in a heterogeneous environment methylene chloride-water in the presence of a phase transfer catalyst have been described. In this embodiment, the main problem lies in the phase transfer. In the embodiment according to EP 506 273 an alkali metal 4-formylphenolate was first isolated and used as the starting material in the subsequent steps of the preparation. The reaction occurred in an anhydrous environment, preferably in ethanol. In EP 816 340 reaction in an anhydrous solvent mixture containing a low-molecular alcohol with another organic solvent (for example toluene) is described.

A drawback of the above-described methods is necessity of pressure reduction of the double bond of compound XI, i.e. 5-(4-(2-(5-ethyl-2-pyridyl)ethoxy)benzylidene)-2,4-thiazolidinedione. The following methods are known:

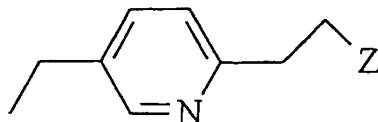
- According to EP 257 781 reduction is carried out with hydrogen under catalysis with rather expensive palladium at the gauge pressure of 50 kg/cm² and the temperature of 50 °C with the yield of 64 %.

- According to EP 506 273 at the gauge pressure of 100 kg/cm² and the temperature of 110 °C the yield is 72 %. Higher yields can be obtained at the expense of increasing the pressure, which entails higher demands for safety of the reaction. Higher temperatures bring about generally higher risk of formation of undesired products.

- An attempt at solving these problems has been given in the patent application WO 93/13095: It relates to reduction with sodium tetrahydroborate catalysed with cobalt chloride.

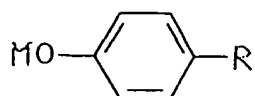
Disclosure of Invention

The solution in accordance with the present invention involves condensing 4-derivatized phenol or phenolate with a pyridine base of general formula III



(III)

wherein Z is a leaving group other than a halogen,
the starting compound of formula II

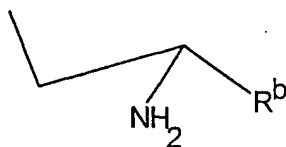


(II)

wherein R is an amino group-containing organic residue, selected from the group comprising a residue of general formula



wherein R^a is hydrogen or a protecting group, which is removed before further treatment,
and a residue of general formula



wherein R^b represents a carboxy group either in the free acid form or in the form of a salt or ester or another functional derivative or the nitrile group CN, and
M represents a hydrogen atom or an alkali metal,

being subjected, before or after carrying out the condensation, to the following operations:

- diazotizing the amino group present in organic residue R
- converting the diazotised residue R into a derivative of 2-halopropionate or 2-halopropionitrile of formula



wherein R^b is as defined above and X is a halogen

- cyclizing the derivative of 2-halopropionate or 2-halopropionitrile with thiourea
- hydrolysing the resulting imine thus giving thiazolidinedione cycle.

Diazotizing

Diazotizing is preformed from the bromohydride of the respective aniline derivative or amino acid by drop-by-drop adding a solution of sodium nitrite in water at about 5 °C.

Converting into the halo derivative of general formula



is performed, depending of the type of the starting compound II, either directly in the course of diazotising in the presence of an excess of halide ions or by adding an acrylate or acrylonitrile and cuprous oxide to the solution of the completed diazonium salt.

Cyclizing with thiourea is carried out in ethanol at its boiling point with subsequent separation of the product.

The subsequent hydrolysis, resulting in formation of the thiazolidinedione cycle, is carried out with hydrochloric acid at its boiling point.

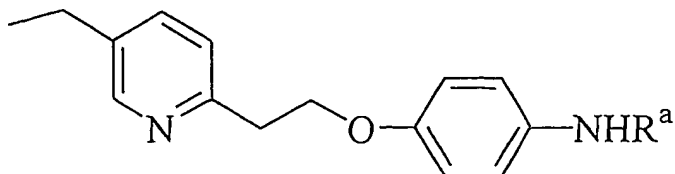
The condensation with 2-(5-ethyl-2-pyridyl)ethyl derivative itself is carried out in the presence of a base in the form of an alkali metal carbonate, hydroxide or hydride in an organic solvent or in a heterogeneous mixture of an organic solvent and water. The temperature when performed in an organic solvent ranges between 50 and 130 °C. After the reaction is complete the solvent can be evaporated, or, if it is a high-boiling solvent, water is added to the mixture. The product is extracted with ethyl acetate.

If R is a residue of formula



wherein R^a is as defined above,

then condensing of compounds II and III results in a compound of formula V

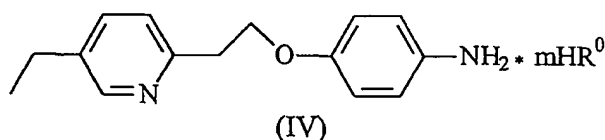


(V)

Where R^a is a leaving group, then formula V represents a yet undisclosed compound. The selection of the leaving group R^a is not essential for further procedure. It can be selected among the groups such as acyl, n-alkyloxycarbonyl, tert-butyloxycarbonyl, benzyloxycarbonyl, 9-fluorenylmethyloxycarbonyl, allyloxycarbonyl, 2-cyanoethoxycarbonyl, dimethylaminomethylenyl or hexa-2,4-dien-2,5-diyl.

Preferably, R^a can be an acyl group derived from lower aliphatic acids (C_1 to C_4), such as acetyl, or from lower aromatic acids, such as benzoyl.

Degradation of the compound of formula V results in a compound of formula IV



wherein m stands for 0 (for a free amine) or 1 (for ammonium salt), R^0 is a mineral or organic acid residue such as a halogen, HSO_4^- , NO_3^- , R^1COO^- or $R^1SO_3^-$, wherein R^1 represents hydrogen or a hydrocarbon residue, which can be advantageously used for the preparation of pioglitazone.

The hydrolysis itself is carried out with a mineral or organic acid of general formula HR^0 or with a base.

In case of basic hydrolysis carried out, e.g., from a mixture of potassium hydroxide and ethanol at the boiling temperature, compound IV represents 4-(2-(5-ethyl-2-pyridyl)ethoxy)aniline, i.e. the free base (m is 0).

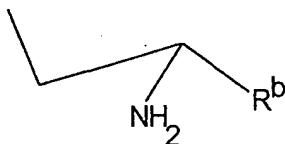
In case of acidic hydrolysis for example haloacids such as hydrochloric acid can be used; compound IV, formed during the reaction, represents 4-(2-(5-ethyl-2-pyridyl)ethoxy)aniline hydrochloride, which, during isolation, is converted into the free base ($m=0$). From the point of view of the subsequent step of pioglitazone synthesis (diazotizing), hydrobromic acid is especially preferable for hydrolysis of compound of formula V, when R^0 represents Br and the product of formula IV, 4-(2-(5-ethyl-2-pyridyl)ethoxy)aniline hydrobromide, need not to be isolated. Oxygen-containing mineral acids such as sulphuric acid can also be used in the hydrolysis. Useful organic acids include alkyl- or arylsulfonic

acids (such as methanesulfonic acid). Numerous carboxylic acids such as formic or acetic acids can be used for the hydrolysis.

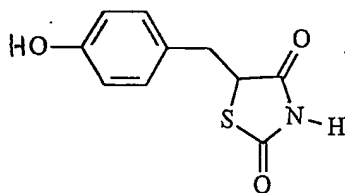
The compound of formula IV is further treated in a known manner, i.e. by diazotizing, Meerwein reaction, cyclizing with thiourea and hydrolysing the imine. This method has been described in EP 193 256 and is here depicted in Scheme 1.

In case of the most common protective group, wherein R^a is acetyl, its decomposition is carried out using a mixture, preferably a mixture of concentrated hydrobromic acid with ethanol in the volume ratio 1: 3.

If R is a residue of formula

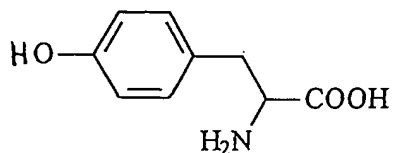


wherein R^b is carboxyl or nitrile, it is preferable to effect the steps of diazotizing, converting into the halo residue, cyclizing a hydrolysing prior to condensation itself. In thus case an intermediate of formula VI



VI

is formed, with which the condensation is subsequently carried out. In this case the starting compound can be the amino acid tyrosine, an inexpensive natural source of formula

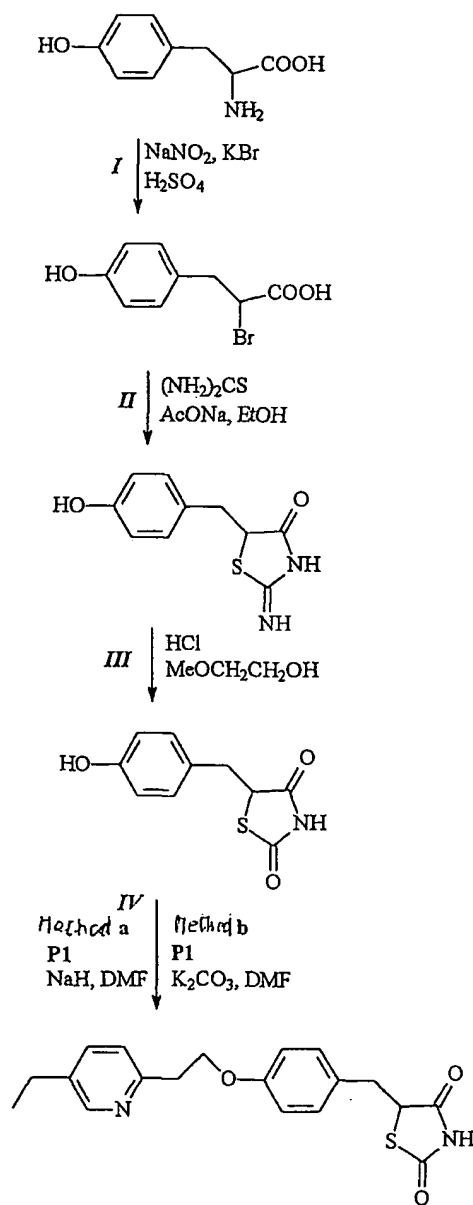


Condensation of intermediate VI with base III can in this case be made in an organic polar solvent such as dimethylformamide or dimethylsulfoxide. The reaction temperature is selected in dependence of the base employed in the range of from 70 to 130 °C. The reaction

time ranges, again in connection with the selected base, between 0.5 and 3 hours. In a usual embodiment, after the reaction is complete the reaction mixture is quenched with water and the product is extracted, usually with acetate.

The complete synthesis of pioglitazone in the above-described special case can be schematically depicted:

Scheme 2

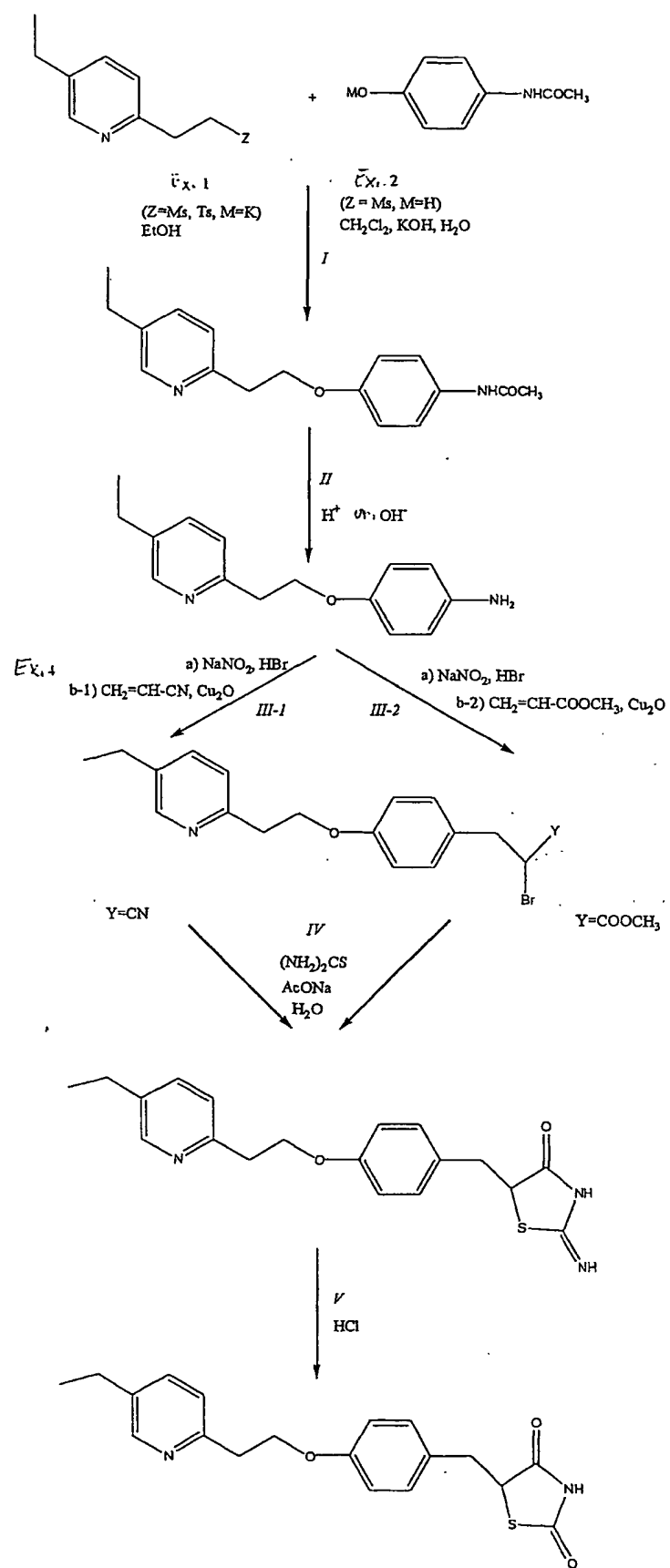


If R is a protected amine of formula -NHR^a , the most preferable composition is the case when R^a is acetyl, and the compound of formula II is widely used paracetamol.

The condensation in this case is preferably carried out in ethanol at 50 °C. After the reaction is complete, ethanol is evaporated and the product extracted with ethyl acetate.

A complete scheme of this preferable procedure with reference to the respective working examples is depicted in Scheme 3:

Scheme 3



Examples

The following examples will further illustrate the present invention.

Reference Example – Preparation of pioglitazone according to EP 193 256

a) Preparation of 4-(2-(5-ethyl-2-pyridyl)ethoxy)nitrobenzene

To a solution of 2-(5-ethyl-2-pyridyl)ethanol (12 g) and 4-fluoronitrobenzene (11.2 g) in dimethylformamide (80 ml) sodium hydride (3.9 g) is added portionwise under ice cooling.

This mixture is agitated under ice cooling for one hour, at room temperature for thirty minutes, poured into ice water and extracted with ethyl acetate. Drying the organic layer and evaporating the solvent gives a yellow solid (4-(2-(5-ethyl-2-pyridyl)ethoxy)nitrobenzene, 21.4 g, 98.2 %). Recrystallizing from a mixture of ether and hexane yields pale yellow crystals (m.p. 45-47 °C). NMR δ (ppm) in CDCl_3 : 1.25 (3H, t, $J=7.6$), 2.65 (2H, q, $J=7.6$), 3.28 (2H, t, $J=6.7$), 4.47 (2H, t, $J=6.7$), 6.95 (2H, m), 7.19 (1H, d, $J=7.9$), 7.49 (1H, dd, $J=2.4$; 8.0), 8.17 (2H, m), 8.41 (1H, d, $J=2.2$).

b) Preparation of 4-(2-(5-ethyl-2-pyridyl)ethoxy)aniline

A solution of 4-(2-(5-ethyl-2-pyridyl)ethoxy)nitrobenzene (23 g) in methanol is hydrogenated under atmospheric pressure in the presence of 10% Pd on carbon (50% humid, 2.5 g). The catalyst is removed by filtering. Distilling the solvent off under reduced pressure gives 4-(2-(5-ethyl-2-pyridyl)ethoxy)aniline as crude oil (20 g, 98 %).

c) Preparation of methyl 2-bromo-3-(4-(2-(5-ethyl-2-pyridyl)ethoxy)phenyl)propionate

To a solution of 4-(2-(5-ethyl-2-pyridyl)ethoxy)aniline (20.4 g) in a mixture of methanol (70 ml) and acetone (180 ml) 47% hydrobromic acid (36 ml) is added. The mixture is cooled. A solution of sodium nitrite (6.12 g) in water (18 ml) is added dropwise at 0 to 5 °C. The mixture is agitated for another twenty minutes at this temperature, then methyl acrylate (43.2 ml) is added and the temperature is elevated to 38 °C. To this mixture cuprous oxide (0.72 g) is added in small portions under mixing. The reaction mixture is mixed as long as nitrogen is released. Then it is concentrated under reduced pressure, the concentrate is alkalified by addition of a concentrated ammonia solution (36 ml) and extracted into ethyl acetate. The organic layer is washed with water and dried over sodium sulfate. Concentrating gives methyl 2-bromo-3-(4-(2-(5-ethyl-2-pyridyl)ethoxy)phenyl)propionate as crude oil (28.4 g; 86.2 %).

NMR δ (ppm) in CDCl_3 : 1.21 (3H, t, $J=7$), 2.60 (2H, q, $J=7$), 3.0-3.6 (4H, m), 3.66 (3H, m), 4.30 (2H, t, $J=7$), 4.3 (1H, m), 6.7-7.5 (6H, m), 8.35 (1H, d, $J=2$)

d) Preparation of 5-(4-(2-(5-ethyl-2-pyridyl)ethoxy)benzyl)-2-imino-4-thiazolidinone

A mixture of methyl 2-bromo-3-(4-(2-(5-ethyl-2-pyridyl)ethoxy)phenyl)propionate (28.4 g), thiourea (5.5 g), sodium acetate (6 g) and ethanol (200 ml) is heated to reflux under heating for three hours. The mixture is concentrated and the concentrate is neutralized with saturated solution of sodium hydrogencarbonate. Adding water (80 ml) and ether (40 ml) precipitates a solid from the mixture. Filtering off gives beige-coloured crystalline

5-(4-(2-(5-ethyl-2-pyridyl)ethoxy)benzyl)-2-imino-4-thiazolidinone (21.4 g; 85.9 %; m.p. 178-182 °C). NMR δ (ppm) in CDCl_3 : 1.2 (3H, t, $J=7.6$), 2.61 (2H, q, $J=7.6$), 2.85 (1H, dd, $J=9.4$; 14.3), 3.14 (2H, t, $J=6.7$), 3.31 (1H, dd, $J=4.1$; 14.2), 4.31 (2H, t, $J=6.7$), 4.52 (1H, dd, $J=4.1$; 9.4), 6.84 (2H, m), 7.14 (2H, m), 7.27 (1H, 6d, $J=7.9$), 7.57 (1H, 6dd, $J=2.3$; 7.9), 8.38 (1H, 6d, $J=2.2$), 8.7 (2H, 6s).

e) Preparation of 5-(4-(2-(5-ethyl-2-pyridyl)ethoxy)benzyl)-2,4-thiazolidindione

A solution of 5-(4-(2-(5-ethyl-2-pyridyl)ethoxy)benzyl)-2-imino-4-thiazolidinone (21.4 g) in 2M hydrochloric acid (200 ml) is heated to reflux for six hours, then concentrated under reduced pressure. The evaporation residue is neutralized with saturated solution of sodium hydrogencarbonate, the crude product is filtered from the mixture. Recrystallizing from a mixture of dimethylformamide and water gives crystalline 5-

4-(2-(5-ethyl-2-pyridyl)ethoxy)benzyl)-2,4-thiazolidinedione (10.6 g; 56.1 %; m.p. 170-173 °C). NMR δ (ppm) in CDCl_3 : 1.24 (3H, t, $J=7.6$), 2.64 (2H, q, $J=7.6$), 3.04 (1H, dd, $J=9.4$; 14.3), 3.21 (2H, t, $J=6.6$), 3.39 (1H, dd, $J=4.0$; 14.2), 4.33 (2H, t, $J=6.6$), 4.48 (1H, dd, $J=4.0$; 9.3), 6.83 (2H, m), 7.12 (2H, m), 7.27 (1H, d, $J=8.0$), 7.50 (1H, dd, $J=2.3$; 8.0), 8.37 (1H, d, $J=1.9$).

Example 1: Preparation of 4-(2-(5-ethyl-2-pyridyl)ethoxy)aniline

Preparation of 4-(2-(5-ethyl-2-pyridyl)ethoxy)acetanilide (condensation – two different starting salts)

a) A mixture of 2-(5-ethyl-2-pyridyl)ethyl mesylate (3.64 g) and potassium 4-acetamidophenolate (3.32 g) in ethanol (18 ml) is agitated at 60 °C for six hours. The mixture

is concentrated and extracted with ethyl acetate and water. The organic layer is washed 3x with a 0.2M solution of sodium hydroxide and 2x with water and dried with sodium sulfate. Evaporating the solvent and washing the residue with ether gives crystalline 4-(2-(5-ethyl-2-pyridyl)ethoxy)acetanilide (2.32 g, 51.3 %; m.p. 85-87 °C).

b) To a solution of 2-(5-ethyl-2-pyridyl)ethyltosylate (15.27 g) in ethanol (50 ml) a solution of potassium 4-acetamidophenolate (9.47 g) in ethanol (40 ml) is added at 50-60 °C during 1.5 hours. The reaction mixture is agitated at the same temperature for another six hours, concentrated and extracted with ethyl acetate (150 ml) and water. The organic layer is washed with a 0.2M solution of sodium hydroxide (4 x 100 ml) and water (2x 100 ml) and dried over sodium sulfate. Evaporating the solvent and washing the residue with ether gives crystalline 4-(2-(5-ethyl-2-pyridyl)ethoxy)acetanilide (6 g, 42.3 %, m.p. 85.5-87 °C) .

Preparation of 4-(2-(5-ethyl-2-pyridyl)ethoxy)aniline (three different modifications)

a) A solution of 4-(2-(5-ethyl-2-pyridyl)ethoxy)acetanilide (13 g) in a mixture of ethanol (80 ml) and concentrated hydrochloric acid (80 ml) is heated to reflux for 2.5 hours. The reaction mixture is neutralized with an aqueous solution of sodium hydroxide and extracted with ethyl acetate. The organic layer is washed with water and dried over magnesium sulfate. Evaporating ethyl acetate gives oily 4-(2-(5-ethyl-2-pyridyl)ethoxy)aniline (9.6 g, 88.1 %).

b) A mixture of 4-(2-(5-ethyl-2-pyridyl)ethoxy)acetanilide (15.5 g), 4 M potassium hydroxide (150 ml) and ethanol (150 ml) is refluxed for 20 hrs and ethanol is evaporated. The mixture is extracted with ethyl acetate; the organic layer is washed with water and dried over sodium sulfate. Evaporating the solvent gives an oily product, 4-(2-(5-ethyl-2-pyridyl)ethoxy)aniline (11.03 g; 80 %)

c) To a solution of 4-(2-(5-ethyl-2-pyridyl)ethoxy)acetanilide (6 g) in ethanol (100 ml) concentrated hydrobromic acid (36 ml) is added. The mixture is heated t reflux for 4 hrs. After evaporating ethanol the residual mixture is used for the next synthetic step without further isolation.

Example 2: 4-(2-(5-ethyl-2-pyridyl)ethoxy)aniline**Preparation of 4-(2-(5-ethyl-2-pyridyl)ethoxy)acetanilide**

To a solution of 2-(5-ethyl-2-pyridyl)ethyl mesylate (1.7 g) and benzyltributylammonium chloride (0.5 g) in dichloromethane (10 ml) a solution of 4-hydroxyacetanilide (1.2 g) and potassium carbonate (1.1 g) in water (10 ml) is added. After agitating for 9 hours and heating to 75 °C the organic layer is separated, washed with water and dried over sodium sulfate.

Evaporating the solvent and washing the residue with ether gives crystalline 4-(2-(5-ethyl-2-pyridyl)ethoxy)acetanilide (1.15 g; 54.8 %; 82-85 °C).

Preparation of 4-(2-(5-ethyl-2-pyridyl)ethoxy)aniline

4-(2-(5-ethyl-2-pyridyl)ethoxy)acetanilide is obtained by hydrolysing 4-(2-(5-ethyl-2-pyridyl)-ethoxy)acetanilide as described in Example 1.

Example 3: Preparation of 5-[[4-[2-(5-ethyl-2-pyridyl)ethoxy]phenyl]methyl]-2,4-thiazolidinedione (pioglitazone)**Preparation of 2-bromo-3-(4-hydroxyphenyl)-propionic acid**

Tyrosine (20 g) is dissolved in HBr (100 ml, 48% diluted with 200 ml water). To the cooled solution NaNO_2 in 30 ml water is added dropwise. The product is extracted with ethyl acetate and after common treatment 2-bromo-3-(4-hydroxyphenyl)-propionic acid is obtained, which is used without purification for the next step.

Preparation of 5-(4-hydroxybenzyl)-1,3-thiazolidine-2,4-dione

2-bromo-3-(4-hydroxyphenyl)-propionic acid is refluxed in ethanol (150 ml) with thiourea (14.7 g) and sodium acetate (16.32 g). After 3.5 hrs the solvent is evaporated. The residue is mixed in water, filtered off and washed with ether. A product (17.15 g) having the melting point 180.5-183.6 °C is obtained.

The obtained 5-(4-hydroxybenzyl)-2-imino-1,3-thiazolidine-4-one (16 g) is dissolved in methoxyethanol (150 ml) and HCl (27 ml) is added. The reaction mixture is refluxed for 4 hours. After evaporating the solvent the residue is mixed with water. The product is extracted from water with ethyl acetate.

The product can be recrystallized from a mixture ethyl acetate-heptane. 9.5 g of 5-(4-hydroxybenzyl)-1,3-thiazolidine-2,4-dione having a melting point 150.9-152.5 °C is obtained.

Preparation of 5-[[4-[2-(5-ethyl-2-pyridyl)ethoxy]phenyl]methyl]-2,4-thiazolidinedione

0.5 g of 5-(4-hydroxybenzyl)-1,3-thiazolidine-2,4-dione is dissolved in ethanol (7 ml) with KOH (0.25 g) and mixed at room temperature for 2 hours. To the obtained salt 2-(5-ethyl-2-pyridyl)ethyl tosylate is added and the reaction mixture is refluxed for 5 hours. After cooling the precipitated salt is filtered off, pH is adjusted and the solvent is evaporated. The semi-solid residue is mixed in water and the solid portion of 5-[[4-[2-(5-ethyl-2-pyridyl)ethoxy]phenyl]methyl]-2,4-thiazolidinedione (0.18 g) is filtered off and recrystallized from the mixture DMF-water.

Example 4: Preparation of 5-[[4-[2-(5-ethyl-2-pyridyl)ethoxy]phenyl]methyl]-2,4-thiazolidinedione (pioglitazone) – another method

Preparation of 2-bromo-3-[4-(2-(5-ethyl-2-pyridyl)ethoxy)phenyl]propionic acid nitrile

To a solution of 4-(2-(5-ethyl-2-pyridyl)ethoxy)phenylammonium bromide (52.8 mmol) in hydrobromic acid (prepared by hydrolyzing 4-(2-(5-ethyl-2-pyridyl)ethoxy)acetanilide according to Example 1c) 25 ml of 48% hydrobromic acid, 50 ml of methanol, 100 ml of acetone are added and the mixture is cooled to 0 °C. A solution of sodium nitrite (4.5 g) in water (10 ml) is added dropwise to the mixture at 0 to 5 °C. The mixture is mixed at 5 °C for another 20 mins, then acrylonitrile (16.8 ml) is added and the temperature is elevated to 38 °C. Cuprous oxide (0.5 g) is added in small portions. The mixture is agitated until release of nitrogen stops, then concentrated under reduced pressure, the concentrate is alkalinized by adding an aqueous ammonia solution and extracted with ethyl acetate. The ethylacetate layer is washed with water and dried over sodium sulfate. Evaporating the solvent gives the nitrile of 2-bromo-3-[4-(2-(5-ethyl-2-pyridyl)ethoxy)phenyl]-propionic acid as crude oil (18.7 g, 99 % of theory).

Preparation of 5-[4-(2-(5-ethyl-2-pyridyl)ethoxy)benzyl]-2-imino-4-thiazolidinone

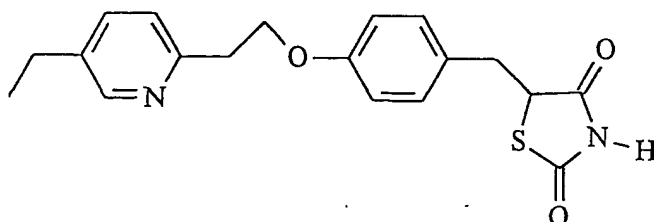
A mixture of 2-bromo-3-[4-(2-(5-ethyl-2-pyridyl)ethoxy)phenyl]propionic acid nitrile (18.7 g), thiourea (3.9 g), sodium acetate (4.2 g) and ethanol (110 ml) is heated to reflux for 6 hours, then it is concentrated under reduced pressure, ethyl acetate (30 ml) and saturated sodium hydrogen carbonate solution (100 ml) are added. The mixture is left to stand

overnight, then filtered and the precipitated beige solid washed with ether to give 5-[4-(2-(5-ethyl-2-pyridyl)ethoxy)benzyl]-2-imino-4-thiazolidinone (11.2 g, 59 % of theory).

5-[4-(2-(5-ethyl-2-pyridyl)ethoxy)benzyl]-2-imino-4-thiazolidinone is then treated in the same manner as in Example 3.

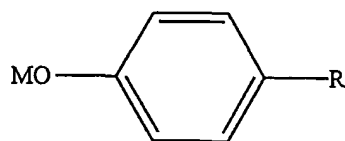
CLAIMS

1. A method for obtaining antidiabetic of formula I



(I)

wherein the method comprises condensing of a 4-derivatized phenol or phenolate of general formula II

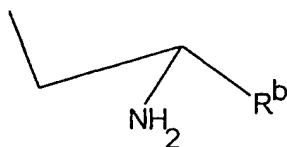


(II)

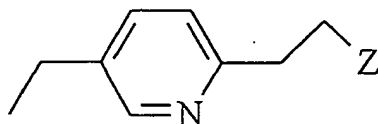
wherein R is an amino group-containing organic residue, selected from the group comprising a residue of general formula



wherein R^a is hydrogen or a protecting group, which is removed before further treatment,
and a residue of general formula



wherein R^b represents a carboxy group either in the free acid form or in the form of a salt or ester or another functional derivative or the nitrile group CN, and
 M represents a hydrogen or alkali metal atom,
 with a pyridine base of general formula III



(III)

wherein Z is a leaving group other than a halogen,
 wherein, before or after carrying out the condensation, the following operations are carried out:

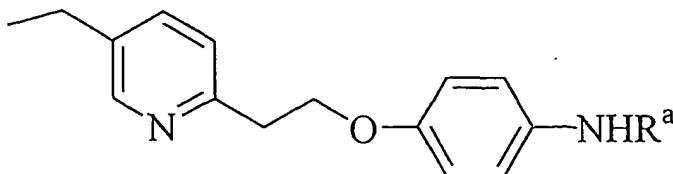
- (a) diazotizing the amino group present in organic residue R
- (b) converting the diazotised residue R into a derivative of 2-halopropionate or 2-halopropionitrile of formula



wherein R^b is as defined above and X is a halogen

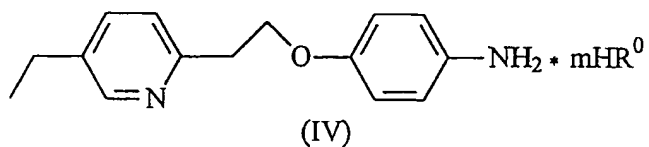
- (c) cyclizing the derivative of 2-halopropionate or 2-halopropionitrile with thiourea
- (d) hydrolysing the resulting imine thus giving pioglitazone of formula I.

2. The method of claim 1 characterized in using the starting compound of formula II in which R is a group of formula $-NHR^a$ wherein R^a is as defined in claim 1 and a starting compound of formula III in which Z is as defined in claim 1 to give a compound of formula V



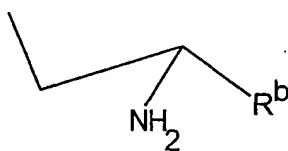
(V)

wherein R^a is as defined above, which is then hydrolyzed into a compound of formula IV

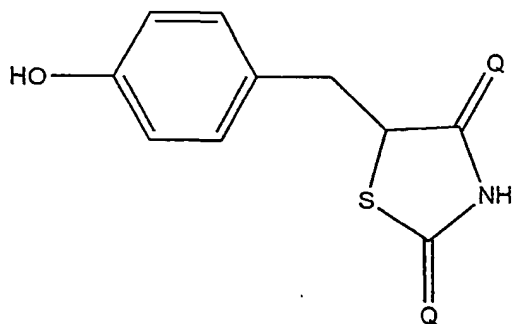


wherein m is 0 (for the free amine) or 1 (for the ammonium salt), R^0 is a residue of a mineral or organic acid such as a halogen, HSO_4^- , NO_3^- , HCO_3^- , R^1COO^- or R^1SO_3^- , wherein R^1 represents hydrogen or alkyl.

3. The method of claim 1 characterized in using a starting compound of formula II wherein R is a group of formula



wherein R^b is as defined in claim 1,
the compound of general formula II being treated with steps (a) to (d) before the condensation and the condensation itself is then carried out with a compound of formula VI

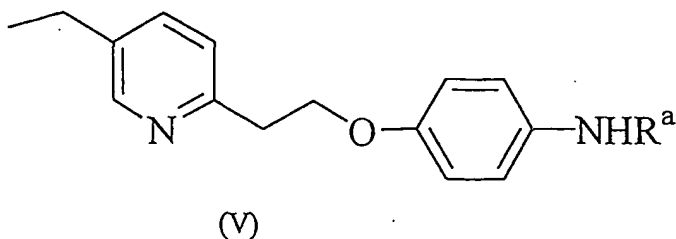


$\text{Q} = \text{O}, \text{NH}$

(VI)

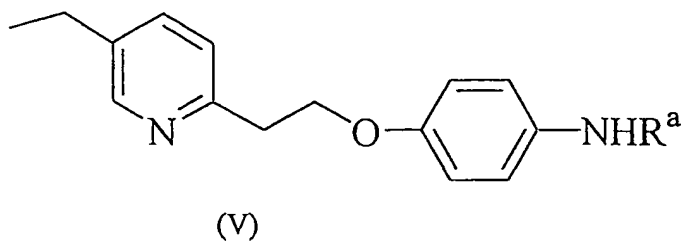
wherein Q represents oxygen or the imino group NH .

4. The method of claim 3 wherein R^b is the nitrile group.
5. The method of claim 1 or 4 wherein Z represents tosyl or mesyl.
6. The method of any of the preceding claims wherein the hydrolysis of the compound of formula V is carried out in an acidic environment.
7. The method of any of the preceding claims wherein the hydrolysis of the compound of formula V is carried out in an alkaline environment.
8. The method of claim 6 wherein the hydrolysis is carried out in a mixture of ethanol and concentrated hydrochloric or hydrobromic acid in a volume ratio of from 3:1 to 1:1 at boiling for 2 to 5 hours.
9. The method of claim 1 or 4 wherein the reaction of compounds II and III is carried out in a heterogeneous mixture of an organic solvent and water in the presence of a phase transfer catalyst.
10. The method of claim 9 wherein the reaction occurs in the mixture of methylene chloride and water in the presence of benzyltributylammonium chloride.
11. The method of claim 1, 3, 4 or 5 wherein the condensation is carried out in a polar organic solvent at 50 to 130 °C in the presence of an alkaline material.
12. A compound of formula V



wherein R^a is a protecting group,
as an intermediate for obtaining the compound of claim 1.

13. The compound of formula V



wherein R^a is the -COCH₃ group.

INTERNATIONAL SEARCH REPORT

Intern Application No
PCT/02/00024

A. CLASSIFICATION OF SUBJECT MATTER

IPC 7 C07D417/12 A61K31/4427 A61P3/10

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 7 C07D A61K A61P

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

EPO-Internal, CHEM ABS Data, BEILSTEIN Data

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	EP 0 193 256 A (TAKEDA CHEMICAL INDUSTRIES LTD) 3 September 1986 (1986-09-03) cited in the application the whole document ---	1-13
A	EP 0 816 340 A (TAKEDA CHEMICAL INDUSTRIES LTD) 7 January 1998 (1998-01-07) cited in the application page 5; claim 1 --- -/--	1

☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex.

* Special categories of cited documents :

A document defining the general state of the art which is not considered to be of particular relevance

E earlier document but published on or after the international filing date

L document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)

O document referring to an oral disclosure, use, exhibition or other means

P document published prior to the international filing date but later than the priority date claimed

T later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

X document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

Y document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art.

& document member of the same patent family

Date of the actual completion of the international search

28 August 2002

Date of mailing of the international search report

10/09/2002

Name and mailing address of the ISA

European Patent Office, P.B. 5818 Patentlaan 2
NL - 2280 HV Rijswijk
Tel. (+31-70) 340-2040, Tx. 31 651 epo nl,
Fax (+31-70) 340-3016

Authorized officer

Bosma, P

INTERNATIONAL SEARCH REPORT

Intern al Application No
PCT/ 2/00024

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	DATABASE CA 'Online! CHEMICAL ABSTRACTS SERVICE, COLUMBUS, OHIO, US; MASUMOTO, KEIICHI ET AL: "Preparation of phenylurea derivatives as herbicides" retrieved from STN Database accession no. 108:75006 XP002211328 abstract; CAS RN 112772-87-9 & JP 62 205054 A (TAKEDA CHEMICAL INDUSTRIES, LTD., JAPAN) 9 September 1987 (1987-09-09) -----	12
X	EP 0 067 718 A (TAKEDA CHEMICAL INDUSTRIES, LTD., JAPAN) 22 December 1982 (1982-12-22) claim 1; examples 2,4 -----	12
X,P	US 2001/008898 A1 (TOMIYAMA, TSUYOSHI ET AL) 19 July 2001 (2001-07-19) compounds 10,15,24 -----	12

INTERNATIONAL SEARCH REPORT

Division on patent family members

Intern. Application No

PCT/02/00024

Patent document cited in search report		Publication date	Patent family member(s)	Publication date
EP 0193256	A	03-09-1986	AT 41931 T AU 572719 B2 BR 1100325 A3 CA 1277323 A1 CS 9104079 A3 DE 3662689 D1 DK 21986 A DK 171614 B1 EP 0193256 A1 ES 550986 D0 ES 8705886 A1 FI 860232 A ,B, GR 860124 A1 HK 3692 A HU 41775 A2 IE 58928 B JP 1853588 C JP 5066956 B JP 61267580 A KR 9210046 B1 LU 90719 A9 LV 5779 A4 MX 9202933 A1 NO 860141 A ,B, PT 81859 A ,B SG 105691 G US 4687777 A ZA 8600203 A	15-04-1989 12-05-1988 27-06-2000 04-12-1990 15-04-1992 11-05-1989 20-07-1986 24-02-1997 03-09-1986 16-05-1987 01-08-1987 20-07-1986 19-05-1986 17-01-1992 28-05-1987 01-12-1993 07-07-1994 22-09-1993 27-11-1986 13-11-1992 26-03-2001 20-12-1996 30-06-1992 21-07-1986 01-02-1986 14-02-1992 18-08-1987 30-09-1987
EP 0816340	A	07-01-1998	CA 2208878 A1 EP 0816340 A1 JP 3256841 B2 JP 10072438 A US 6100403 A US 5952509 A	27-12-1997 07-01-1998 18-02-2002 17-03-1998 08-08-2000 14-09-1999
JP 62205054	A	09-09-1987	NONE	
EP 0067718	A	22-12-1982	JP 57209274 A BR 8203448 A EP 0067718 A1 US 4443246 A	22-12-1982 07-06-1983 22-12-1982 17-04-1984
US 2001008898	A1	19-07-2001	JP 2001261662 A CN 1315320 A DE 10100772 A1 FR 2804681 A1 GB 2359082 A	26-09-2001 03-10-2001 19-07-2001 10-08-2001 15-08-2001

(19) World Intellectual Property
Organization
International Bureau



(43) International Publication Date
7 November 2002 (07.11.2002)

PCT

(10) International Publication Number
WO 2002/088120 A1

(51) International Patent Classification⁷: C07D 417/12,
A61K 31/4427, A61P 3/10

(21) International Application Number:
PCT/CZ2002/000024

(22) International Filing Date: 25 April 2002 (25.04.2002)

(25) Filing Language: English

(26) Publication Language: English

(30) Priority Data:
PV 2001-1501 26 April 2001 (26.04.2001) FR
PV 2001-1502 26 April 2001 (26.04.2001) CZ

(71) Applicant (for all designated States except US): LÉCIVA,
A.S. [CZ/CZ]; Dolní Mecholupy 130, 102 37 Praha 10
(CZ).

(72) Inventors; and

(75) Inventors/Applicants (for US only): HALAMA, Ales

[CZ/CZ]; Lonkova 481, 530 09 Pardubice (CZ). HEJTMÁNKOVÁ, Ludmila [CZ/CZ]; Benesovice 13, 565 01 Přelouč (CZ). LUSTIG, Petr [CZ/CZ]; Husova 255, 530 03 Pardubice (CZ). RICHTER, Jindřich [CZ/CZ]; Hurka 255, 530 12 Pardubice (CZ). SRSNOVÁ, Lucie [CZ/CZ]; Čs. Armády 471, 537 01 Chrudim (CZ). JIRMAN, Josef [CZ/CZ]; Ceskova 1706, 530 02 Pardubice (CZ).

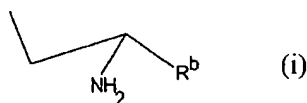
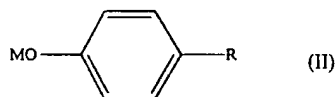
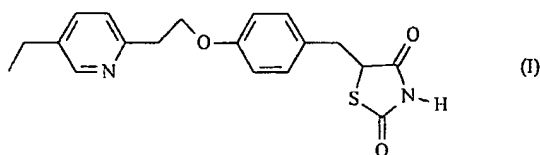
(74) Agent: JIROTKOVÁ, Ivana; Rott, Ruzicka & Guttman,
P.O. Box 94, Nad Stolou 12, 170 00 Praha 7 (CZ).

(81) Designated States (national): AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VN, YU, ZA, ZM, ZW.

(84) Designated States (regional): ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent (AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR,

[Continued on next page]

(54) Title: METHOD FOR OBTAINING PIOGLITAZONE AS AN ANTIDIABETIC AGENT



(57) Abstract: A method for obtaining antidiabetic of formula (I), wherein the method comprises condensing of a 4-derivatized phenol or phenolate of general formula (II), wherein R is an amino group-containing organic residue, selected from the group comprising a residue of the following formula -NHR³, wherein R³ is hydrogen or a protecting group, which is removed before further treatment, and a residue of general formula (A), wherein R^b represents a carboxy group either in the free acid form or in the form of a salt or ester or another functional derivative or the nitrile group CN, and M represents a hydrogen or alkali metal atom, with a pyridine base of general formula (III), wherein Z is a leaving group other than a halogen, wherein, before or after carrying out the condensation, the following operations are carried out: (a) diazotizing the amino group present in organic residue R; (b) converting the diazotised residue R into a derivative of 2-halopropionate or 2-halopropionitrile of formula (B), wherein R^b is as defined above and X is a halogen; (c) cyclizing the derivative of 2-halopropionate or 2-halopropionitrile with thiourea; (d) hydrolysing the resulting imine thus giving pioglitazone of formula (I).



GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG).

(AM, AZ, BY, KG, KZ, MD, RU, TJ, TM), European patent (AT, BE, CH, CY, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE, TR), OAPI patent (BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG)

Declarations under Rule 4.17:

- as to applicant's entitlement to apply for and be granted a patent (Rule 4.17(ii)) for the following designations AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NO, NZ, OM, PH, PL, PT, RO, RU, SD, SE, SG, SI, SK, SL, TJ, TM, TN, TR, TT, TZ, UA, UG, UZ, VN, YU, ZA, ZM, ZW, ARIPO patent (GH, GM, KE, LS, MW, MZ, SD, SL, SZ, TZ, UG, ZM, ZW), Eurasian patent

- of inventorship (Rule 4.17(iv)) for US only

Published:

- with international search report
- with amended claims

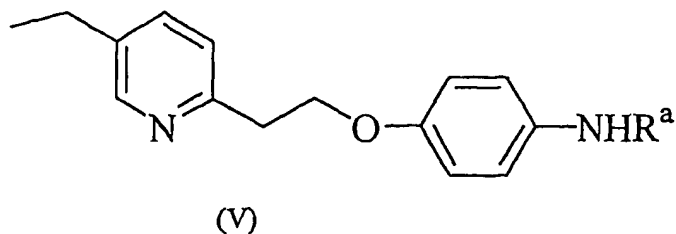
Date of publication of the amended claims: 26 February 2004

For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

AMENDED CLAIMS

[received by the International Bureau on 05 November 2002 (05.11.02);
original claim 12 replaced by amended claim 12 (2 pages)]

4. The method of claim 3 wherein R^b is the nitrile group.
5. The method of claim 1 or 4 wherein Z represents tosyl or mesyl.
6. The method of any of the preceding claims wherein the hydrolysis of the compound of formula V is carried out in an acidic environment.
7. The method of any of the preceding claims wherein the hydrolysis of the compound of formula V is carried out in an alkaline environment.
8. The method of claim 6 wherein the hydrolysis is carried out in a mixture of ethanol and concentrated hydrochloric or hydrobromic acid in a volume ratio of from 3:1 to 1:1 at boiling for 2 to 5 hours.
9. The method of claim 1 or 4 wherein the reaction of compounds II and III is carried out in a heterogeneous mixture of an organic solvent and water in the presence of a phase transfer catalyst.
10. The method of claim 9 wherein the reaction occurs in the mixture of methylene chloride and water in the presence of benzyltributylammonium chloride.
11. The method of claim 1, 3, 4 or 5 wherein the condensation is carried out in a polar organic solvent at 50 to 130 °C in the presence of an alkaline material.
12. A compound of formula V



wherein R^a is a protecting group selected among the groups acyl, n-alkyloxycarbonyl, tert-butyloxycarbonyl, benzyloxycarbonyl, 9-fluorenylmethyloxycarbonyl,

allyloxycarbonyl, 2-cyanoethoxycarbonyl, dimethylaminomethylenyl or hexa-2,4-dien-2,5-diyl,

as an intermediate for obtaining the compound of claim 1.